

FACULTY OF SCIENCES DEPARTMENT OF LIFE SCIENCES

COURSE: B.Sc. SUBJECT NAME: Biochemistry Teaching & Evaluation Scheme:-

SEMESTER: II SUBJECT CODE: 4SC02BIC1

Tea	ching	hour	s/week	Credit	Evaluation Scheme/semester							
						Theory Practic					ctical	
Th	Tu	Pr	Total			Sessional University Exam Exam Internal U		University	Total Marks			
					Marks	Hrs	Marks	Hrs	Pr	TW		
4	0	4	8	6	30	1	70	3	30		70	200

Objectives:-The objective of this course is that the students can learn about basics of Biochemistry.

Prerequisites:-Basic knowledge of Chemistry and Biological science.

Sr.	Course Contents	Hours
No.		
1	Bioenergetics: First and second laws of Thermodynamics, Definitions of Gibb's Free Energy, enthalpy and Entropy and mathematical relationship among them, Standard free energy change and equilibrium constant Coupled reactions and additive nature of standard free energy change, Energy rich compounds: Phosphoenol pyruvate, 1,3-Bisphosphoglycerate, Thioesters, ATP	10
2	Carbohydrates:	12
	Families of monosaccharides: aldosesandketoses, trioses, tetroses, pentoses, and hexoses, Stereo isomerism of monosaccharides, epimers, Mutarotation and anomers of glucose. Furanoseand pyranose forms of glucose and fructose, Haworth projection formulae for glucose; chair and boat forms of glucose, Sugar derivatives, glucosamine, galactosamine, muramic acid,N-acetyl neuraminic acid, Disaccharides; concept of reducing and non-reducing sugars, occurrence and Haworth projections of maltose, lactose, and sucrose, Polysaccharides, storage polysaccharides, starch and glycogen. Structural Polysaccharides, cellulose, peptide glycan and chitin	
3	Lipids: Definition and major classes of storage and structural lipids. Storage lipids. Fatty acids structure and functions. Essential fatty acids. Triacyl glycerols structure, functions and properties. Saponification Structural lipids. Phospho glycerides: Building blocks, General structure, functions and properties. Structure of phosphatidyl ethanol amine and phosphatidyl choline, Sphingo lipids: building blocks, structure of sphingosine, ceramide. Special mention of sphingomyelins, cerebrosides and gangliosides Lipidfunctions: cell signals, cofactors, prostagl and ins, Introduction of lipid micelles, monolayers, bilayers	12





4	Proteins:	10
	Functions of proteins, Primary structures of proteins: Amino acids, the	
	building blocks of proteins. General formula of amino acid and concept of	
	zwitter ion. Titration curve of amino acid and its Significance, Classification,	
	biochemical structure and notation of standard protein amino acids Ninhydrin reaction. Natural modifications of amino acids in protein	
	shydrolysine, cystineand hydroxyproline, Non protein amino acids:	
	Gramicidin, beta-alanine, D-alanine and D-glutamicacid Oligopeptides:	
	Structure and functions of naturally occurring glutathione and insulin and	
	synthetic aspartame, Secondary structure of proteins: Peptide unit and its	
	salient features. The alpha helix, the beta pleated sheet and their occurrence	
	in proteins, Tertiary and quaternary structures of proteins. Force sholding the	
	polypeptide together. Human haemoglobin structure,Quaternarystructuresof	
5	proteins.	
5	Enzymes: Structure of any main Anony main and actactors prosthatiography TDD	
	Structure of enzyme: Apoenzyme and cofactors, prostheticgroup-TPP,	
	coenzyme NAD, metalcofactors, Classification of enzymes, Mechanism of	
	action of enzymes: activesite, transition state complex and activation energy.	
	Lock and key hypothesis, and Induced Fithypothesis. Significance of	10
	hyperbolic, double reciprocal plots of enzyme activity, Km, and allosteric	
	mechanism Definitions of terms–enzyme unit, specific activity and turnover	
	number, Multienzyme complex: pyruvate dehydrogenase; isozyme: lactate	
	dehydrogenase, Effect of pH and temperature on enzyme activity. Enzyme	
	inhibition: competitive-sulfadrugs; non-competitive-heavy metal salts	
6	Vitamins:	
	Classification and characteristics with suitable examples, sources and	06
	importance.	
	Total	60

Learning Outcomes:- The students are expected to

- Understand the classification of Carbohydrate, lipid, protein, Enzymes and Vitamins.
- Understand the basic structure and function of various Biochemicals.

- 1. Campbell, MK(2012)Biochemistry,7thed.,PublishedbyCengageLearning
- **2.** Campbell, PN and SmithAD (2011) Biochemistry Illustrated, 4thed., Published by Church illLivingstone
- **3.** T ymoczko JL, Berg JM and StryerL (2012) Biochemistry: Ashortcourse, 2nded., W.H.Freeman
- 4. Berg JM, T ymoczko JL and StryerL (2011) Biochemistry, W.H. Freeman and Company
- 5. Nelson DL and CoxMM (2008) Lehninger Principles of Biochemistry, 5thEdition., W.H.Freeman and Company
- 6. WilleyMJ,Sherwood,LM&WoolvertonCJ(2013)Prescott,HarleyandKlein'sMicrobi ology by.9thEd.,McGrawHill



7. Voet, D. and Voet J.G (2004) Biochemistry 3rd edition, John Wiley and Sons,

Sr. No.	Course Contents								
1	Properties of water, Concept of pH and buffers, preparation of buffers and								
	Numerical problems to explain the concepts								
2	Numerical problems on calculations of Standard Free Energy Change and								
	Equilibrium constant								
3	Standard Free Energy Change of coupled reactions								
4	Qualitative/Quantitative tests for carbohydrates, reducing sugars, non reducing								
	sugars								
5	Qualitative/Quantitative tests for lipids and proteins								
6	Study of protein secondary and tertiary structures with the help of models								
7	Study of enzyme kinetics– calculation of V_{max} , Km, Kcat values								
8	Study effect of temperature on enzyme activity								
9	Study effect of temperature pH on enzyme activity								
10	Estimation of any one vitamin								





FACULTY OF LIFE SCIENCES DEPARTMENT OF MICROBIOLOGY

COURSE: B.Sc. SUBJECT NAME: Molecular Biology Teaching & Evaluation Scheme:-

SEMESTER: II SUBJECT CODE: 4SC02MOB1

Tea	ching	hour	s/week	Credit	Evaluation Scheme/semester							
					Theory Practic						ctical	
Th	Tu	Pr	Total		Sessio Exar		Univer Exai	·	Inte	ernal	University	Total Marks
					Marks	Hrs	Marks	Hrs	Pr	TW		
4	0	4	8	6	30	1	70	3	30		70	200

Objectives:-The objective of this course is that the students can learn about Molecular Biology.

Prerequisites:-Basic knowledge of Biological Sciences.

Sr.	Course Contents	Hours
No.		
1	Structures of DNA and RNA/Genetic Material :	
	DNA Structure: MieschertoWatson and Crick-historic perspective,	
	DNAstructure, Salient features of double helix, Types of DNA, Types of	
	genetic material, Denaturation and renaturation, cotcurves. DNA topology-	12
	linking number, topoisomerases; Organization of DNA Prokaryotes,	
	Viruses, Eukaryotes. RNA Structure, Organelle DNA- mitochondria and	
	chloroplast DNA.	
2	Replication of DNA(ProkaryotesandEukaryotes):	
	Bidirectional and unidirectional replication, semi-conservative, semi-	
	discontinuous replication Mechanism of DNAreplication: Enzymes and	
	proteins involved in DNA replication- DNA polymerases, DNAligase,	10
	primase, telomerase- for replication of linearends, Various models of DNA	
	replication including rolling circle, D-loop(mitochondrial), Θ (theta)mode of	
	replication and other accessory protein, Mismatch and excision repair	
3	Transcription in Prokaryotes and Eukaryotes:	
	Transcription: Definition, difference from replication, promoter-concept and	
	strength of promoter, RNA Polymerase and the transcription unit,	8
	Transcription in Eukaryotes: RNA polymerases, general Transcription	
	factors	
4	Post-Transcriptional Processing:	
	Splitgenes, concept of introns and exons, RNA splicing, spliceosome	8
	machinery, concept of alternative splicing, Polyadenylation and capping,	0
	Processing of rRNA, RNA interference:siRNA, miRNA and its significance	





5	Translation (Prokaryotes and Eukaryotes): Translational machinery, Charging of tRNA, aminoacyl tRNA synthetases, Mechanisms of initiation, elongation and terminate on of polypeptide in both prokaryotes and eukaryotes, Fidel it of translation, Inhibitors of protein synthesis in prokaryotes and eukaryote	10
6	Regulation of gene Expression in Prokaryotes and Eukaryotes: Principles of transcriptional regulation, regulation at initiation with examples from <i>lac</i> and <i>trp</i> operons, Sporulation in <i>Bacillus</i> , Yeast mating types witching, Changes in Chromatin Structure- DNA methylation and Histone Acetylation mechanisms.	12
	Total	60

Learning Outcomes:-At the end of the course the student would have sufficient knowledge of Introduction to Microbiology.

- Watson JD, Baker TA, Bell SP, Gann A, Levine M and Losick R (2008) Molecular Biology of the Gene, 6^a edition, Cold Spring Harbour Lab. Press, Pearson Publication
- 2) Becker WM, Kleinsmith LJ, Hardin J and Bertoni GP (2009) The World of the Cell, 7thedition, Pearson Benjamin Cummings Publishing, SanFrancisco
- 3) **De Robertis EDP and De Robertis EMF** (2006) Cell and Molecular Biology, 8thedition. Lippincott Williams and Wilkins, Philadelphia
- 4) **KarpG** (2010) Celland Molecular Biology: Concepts and Experiments, 6thedition, John Wiley&Sons.Inc.
- 5) **Sambrook J and Russell DW**. (2001). Molecular Cloning: A Laboratory Manual.4th Edition, Cold Spring Harbour Laboratorypress.
- 6) Krebs J, Goldstein E, Kilpatrick S (2013). Lewin's Essential Genes, 3rdEd., Jones and Bartlett Learning





Sr.	Course Contents
No.	
1	Study of different types of DNA and RNA using micro graphs and
	model/schematic representations
2	Study of semi-conservative replication of DNA through
	micrographs/schematic representations
3	Isolation of genomic DNA from <i>E.coli</i>
4	Estimation of salmon sperm/calfthymus DNA using colorimeter (diphenylamine
	reagent) or UV spectrophotometer (A260measurement)
5	Estimation of RNA using colorimeter (orcinol reagent) or UV spectrophotometer
	(A260measurement).
6	Resolution and visualization of DNA by Agarose Gel Electrophoresis.
7	Resolution and visualization of proteins by Polyacrylamide Gel Electrophoresis (SDS-
	PAGE).







FACULTY OF LIFE SCIENCES DEPARTMENT OF MICROBIOLOGY

COURSE: B.Sc. SUBJECT NAME: Computational Skills 4SC02COS2

SEMESTER: II SUBJECT CODE:

Teaching & Evaluation Scheme:-

Tea	ching	hour	s/week	Credit		Evaluation Scheme/semester						
					Theory Pr						ctical	
Th	Tu	Pr	Total					University Exam Internal		University	Total Marks	
					Marks	Hrs	Marks	Hrs	Pr	TW		
2	0	2	4	3	20	1	50	2			30	100

Objectives:-Computer knowledge is very essential in each and every department. The course aims to provide ample computer knowledge to students in such a manner that it is useful in their future studies.

Prerequisites:-Student should have basic knowledge of computers, which would enable them to learn the course in a more effective manner.

Sr.	Course Contents	Hours
No.		
1	Basic fundamentals of computer.	04
	Definition, characteristics, history, computer terminology, computer	
	organization, input & output devices, storage devices (including latest	
	devices), classifications of computers (including current computer systems),	
	application of computers in lifescience, introduction to computer virus,	
	problems associated with virus infection and its remedies	
2	Windows Basic introduction.	04
	Operating Systems	
	Definition, functions of an operating system, types of operating systems and	
	their characteristics, difference between operating system and application	
	Software.	
	Basic Dos commands both internal and external.	





	Of WAR T	
3	MS Word:	09
	Word Essentials, the word workplace, Parts of MS Word screen, Typing and	
	Editing, Finding and Replacing, Autocorrect and Auto text, Reusing Text and	
	Graphics, use of spell-check & grammar, thesaurus and scientific symbols,	
	viewing of document by various ways Editing Tools, Formatting Text	
	Formatting Text Character, Formatting Paragraphs, Formatting and Sorting	
	Lists, Page Design and Layout, Page Setup : Margins, Page Numbers, and	
	Other Items, Newspaper -style Columns, Working with Tables Creating and	
	formatting of tables and sorting, merging etc. of data in tables. Inserting,	
	deleting and sizing of rows and columns in tables, Opening, Saving and	
	Protecting Documents, locating and Managing Documents Printing,	
	Assembling Documents with Mail Merge, references.	
4	Ms Excel:	09
	Introduction to EXCEL worksheet, calculations in EXCEL. Hierarchy of	
	operation, library functions such as logarithm, square root, standard deviation,	
	sum, average, t-test, ANOVA etc. Drawing graphs in EXCEL line graph,	
	histogram, pie-chart.	
	-Editing chart features such as annotation, labeling of axis, changing legends	
	etc.	
5	MS PowerPoint	04
	Creating and viewing a presentation, adding animations and managing slides	
	etc	
6	Networking & Internet:	05
	Computer networks, networking technology, components of network. Internet	
	- Basic terms, software and hardware requirement for internet, process of	
	internetworking, internet tools, Email- components and working, study of	
	biotechnological/microbiological/biochemistry web sites and search engines,	
	searching through data bases, study of patent websites.	
7	Introduction to the following software	04
	MS Paint, MS Access, Outlook, Adobe acrobat reader, Adobe Professional,	
0	Chemdraw, ISIS Draw, Nero Burning ROM.	07
8	Basic Biostatistics	06
	Introduction, Mean, Median; Standard error, Standard deviation, Variance.	

Learning Outcomes:-The course aims at providing the students ample knowledge of computational skills which shall be of use in their academic as well as professional life.

- 1. **Taxali R.K., P.C**. Software for Windows 98 made simple 8th Edition 2002 Tata Mc, New Delhi.
- 2. Guy Hart Davis, WORD 2000, BPB Publications, New Delhi, 1999
- 3. Joyce Cox, MS Office: Step by Step, Prentice Hall of India, New Delhi, 2007
 - **4.** Cornell, Accessing and Analysing Data with MS EXCEL, Prentice Hall of India, New Delhi, 20





FACULTY OF LIFE SCIENCES

DEPARTMENT OF MICROBIOLOGY

COURSE: **B.Sc.**

SEMESTER: II

SUBJECT NAME: Generic Elective-II (Basics of Virology) SUBJECT CODE: 4SC02BAV1

Teaching & Evaluation Scheme:-

Teaching hours/week Cre					Evaluation Scheme/semester							
	Theory					ory Pract			octical			
Th	Tu	Pr	Total		Sessional Exam		University Exam		y Internal		University	Total Marks
					Marks	Hrs	Marks	Hrs	Pr	TW		
4	0	0	4	4	30	1	70	3				100

Objectives:-The objective of this course is that the students can learn about the viruses.

Prerequisites:-Basic knowledge of Biological Sciences.

Course outline:-

Cours	se outline:-	
Sr. No.	Course Contents	Hours
1	Introduction to Viruses: Properties of viruses; general nature and important features Subviral particles; viroids, prions and their importance Isolation and cultivation of viruses	8
2	Structure, and multiplication of viruses: Morphological characters: Capsid symmetry and different shapes of viruses with examples Viral multiplication in the Cell: Lytic and lysogenic cycle Description of important viruses: salient features of the viruses infecting different hosts Bacteriophages (T4Λ); Plant (TMV& Cauliflower Mosaic Virus), Human (HIV & Hepatitis viruses)	12
3	Role of Viruses in Disease and its prevention Viruses as pathogens: Role of viruses in causing diseases Prevention and control of viruses: Viral vaccines, interferons and anti-viral compounds	10
	Total	30

Learning Outcomes:-At the end of the course the student would have sufficient knowledge of viruses, its structures and pathogenicity of viruses.

Books Recommended:-

- 1. AtlasRM. (1997). Principles of Microbiology. 2nd edition.WM.T. Brown Publishers
- 2. Madig and MT, Martinko JM, Dunlap PV and Clark DP (2014). Brock Biology of Micro-organisms. 14thedition. Pearson Education, Inc.
- 3. Stanier RY, Ingraham JL, Wheelis ML and Painter PR. (2005). General Microbiology.5th edition. McMillan

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- 4. Carter J and Saunders V (2007). Virology; principles and Applications. John Wiley and Sons
- 5. Flint S J, Enquist, L W, Krug, RM, Racaniello, V R Skalka, AM (2004) Principles of Virology, Molecular Biology, Pathogenesis and Control.2nd edition. ASMPress
- 6. Shors Teri (2013) Understanding Viruses2nd edition one sand Bartlett Learning
- Burlington USA
 7. Pelczar Jr MJ, Chan ECS, and Krieg NR. (2004). Microbiology.5th edition Tata Mc Graw Hill.
- 8. Tortora GJ, Funke BR, and Case CL. (2008) Microbiology: An Introduction. 9 edition Pears on Education.



Sr. No.	Course Contents							
1	Study the morphological structures of viruses (DNAandRNA) and their important							
	characters using electron micrographs							
2	Study of the methods of isolation and propagation of plant viruses							
3	Study of cytopathic effects of viruses using photographs							





FACULTY OF SCIENCES DEPARTMENT OF CHEMISTRY

COURSE: B.Sc.

SEMESTER: II

SUBJECT NAME: Stereochemistry in organic synthesis SUBJECT CODE: 4SC02SOS1 Teaching & Evaluation Scheme:-

Teaching hours/week Credit **Evaluation Scheme/semester** Theory Practical Sessional University Total Th Tu Pr Total Internal University Marks Exam Exam Marks Hrs Marks Hrs Pr TW 2 0 2 4 3 50 30 100 20 2 --1 ---

Objectives:-The course will help the student to understand the basic mechanism behind various organic reactions.

To assist students with basic knowledge of stereochemistry and explain how stereochemistry of a compound can affect reaction mechanism.

To understand characteristic and application of chemical Kinetics.

Prerequisites:-Students should have basic knowledge of organic chemistry.

Sr.	Course Contents					
No.						
1	General treatment of reaction mechanisms Ionic and radical reactions; heterolytic and, homolytic bond cleavage Reactive intermediates: carbocations (carbenium and carbonium ions), carbanions, carbon radicals, carbenes – structure using orbital picture, electrophilic/nucleophilic behaviour, stability, generation and fate. Reaction kinetics: transition state theory, rate constant and free energy of activation, free energy profiles for one step and two step reactions, Nucleophilic substitution reactions- SN1, SN2, SNi mechanisms. Effect of substrate structure, nucleophiles and medium on reactivity and mechanism; neighboring group participations. Elimination Reactions- E1, E2, and E1cB mechanisms. Saytzeff and Hofmann rules. Elimination vs substitution reaction. Electrophilic and Activated Nucleophilic substitution reactions of Benzene (Nitration, sulphonation, Halogenation and Friedel Craft reactions)	10				
2	Bonding and Stereochemistry of Carbon Compounds Concept of hybridisation, resonance (including hyperconjugation), inductive effect Huckel's rules for aromaticity & antiaromaticity. bond distance, bond angles Tautomerism: keto-enol tautomerism, Ionization of acids and bases: effect of structure, substituent and solvent on acidity and basicity.(Simple Aliphatic and aromatic Acids, Phenols and amines)	10				





3 Stereochemistry Optical activity of chiral compounds: specific rotation, racemisation (general principle), resolution of simple acids and bases, Representation of molecules in saw horse, Fischer, flying-wedge and Newman formulae and their inter translations, Configuration: stereo centres: systems involving 1, 2, 3 centres, stereogenicity, chirotopicity. Pseudo asymmetric 10 (D/L and R/S descriptor threo/erythro and syn/anti nomenclatures ii) stereo axis in C=C & C=N systems, cis/trans, syn/anti, E/Z descriptors. Stereo selective and stereo specific reactions, chiral reagents, stereochemistry of biphenyls, allenes, and spirans – specification of their configuration.

Learning Outcomes:- This course is designed to

- Deliver a detailed understanding of different types of bonding which are responsible for formation of compounds.
- Apply stereochemistry. It also makes them understand the impacts of stereochemistry in reaction.

- 1. P. Sykes. A Guide to Organic Reaction Mechanism.
- 2. Arun Bahl and B. S. Bahl, Advanced Organic Chemistry- S. Chand.
- 3. S. Sengupta, Basic Stereochemistry of Organic Compounds.
- 4. I.L.Finar. Organic Chemistry (vol.1&2).
- 5. R. T. Morrison & R. N. Boyd:Organic Chemistry, Prentice Hall.
- 6. D. Nasipuri. Stereochemistry of Carbon Compounds.
- 7. E. L. Eliel, Stereochemistry of Carbon Compounds- Tata McGraw Hill.
- 8. T. W. Graham Solomons: Organic Chemistry, John Wiley and Sons.





Sr. No.	Course Contents
1	Determination of viscosity
2	Determination of effect of temperature on viscosity.
3	Determination of effect of concentration on viscosity.
4	Determination of surface tension of given liquid (Drop weight).
5	Determination of surface tension of given liquid (Drop Count).
6	Study of effect of concentration on Surface tension.
7	Determination of the Adsorption Isotherm for adsorption of acetic acid on charcoal.
8	Determination of the order and rate constant of reaction of acid catalyzed hydrolysis of ethyl acetate
9	To determine the partition coefficient of iodine between carbon tetra chloride (CCl ₄) and distilled water.
10	To determine the partition coefficient of benzoic acid between benzene and distilled water.
11	Synthesis of Phthalic anhydride from Phthalic acid.
12	Synthesis of Methyl salicylate from Salicylic acid.
13	Synthesis of Naroline from β-naphthol.
14	Synthesis of Phthalimide from Phthalic anhydride.



FACULTY OF SCIENCES DEPARTMENT OF ENGLISH

SEMESTER: II

COURSE: B.Sc. SUBJECT NAME: Functional English-II SUBJECT CODE: 4SC02FUE1 Teaching & Evaluation Scheme:-

Tea	aching	hour	s/week	Credit	Evaluation Scheme/semester							
]	Theory				Practical			
Th	Tu	Pr	Total	Sessional University Exam Exam		Inte	ernal	University	Total Marks			
					Marks	Hrs	Marks	Hrs	Pr	TW		
2	0	0	2	2	30	1	70	3				100

Objectives:-

- To train students in/for basic fundamentals skills of Communication LSRW in English.
- To provide them the value education for better society.
- To make them able to communicate well in the Professional world.

Prerequisites:-

- Students should have basic knowledge of English Language and grammar.
- Students should have ability to speak and write correct sentences in their day to day language.
- Students should be familiar with correct usage of language.

Sr.	Course Contents	Hours
No.		
	Section: A Linguistic Proficiencies	T
01	Types of Sentences Affirmative, Interrogative, Imperative, Exclamatory	04
02	Modal Auxiliaries (Shall, Will, Can, Could, May, Might, Must, Should, Would and Ought to)	04
03	Causal Constructions: (Get, Make and Have)	01
04	Degrees of Comparison	03
05	Voices (Active Passive)	03
06	Speech (Direct to Indirect)	04
07	Use of Punctuations	03
08	Section: B Literature Part-1 Prose:- 1) Too Dear- Leo Tolstoy 2) An Astrologer's Day -R. K. Narayan 3) A Gift of Magi -O' Henry Part-2 Poetry:- 1) Photographing Mother -Sundram 2) All the World's a Stage -William Shakespeare 3) Stopping By Woods on a Snowy Evening -Robert Frost	08



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Learning Outcomes:-

- The students are expected to understand the basics of English Language and grammar.
- The students should be able to communicate well.

Resources:

- 1. An Intermediate English Grammar, Raymond Murphy, Cambridge University Press.
- 2. A High School English Grammar, Wren & Martin, S. Chand Publication.
- 3. Contemporary English Grammar Structures & Composition, David Green, Macmillan Publishers India.
- 4. Contemporary Indian Short Stories, Series I & II, Sahitya Academy, New Delhi.
- 5. *Modern Gujarati Poetry: A Selection*, translated by **SagunaRamnathan and Rita Kothari**, Sahitya Academy (English Translation), New Delhi.
- **6.** *Effusions: An Anthology of English Prose and Poetry*, ed. by Marathwada University, Oxford University Press, 1987.